

Dichloro-*trans*-bis[tris(2-cyanoethyl)phosphine]-palladium(II)

Simon E. Lawrence,* Maureen Kelly and Orla M. Ní Dhubhghaill

Department of Chemistry, University College Cork, Cork, Ireland

Correspondence e-mail: s.lawrence@ucc.ie

Key indicators

Single-crystal X-ray study

 $T = 293\text{ K}$ Mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$ R factor = 0.027 wR factor = 0.076

Data-to-parameter ratio = 17.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, dichloro-*trans*-bis[tris(2-cyanoethyl)phosphine]palladium(II), $[\text{PdCl}_2(\text{C}_9\text{H}_{12}\text{N}_3\text{P})_2]$, was obtained by the addition of tris(2-cyanoethyl)phosphine to dichloro-*trans*-bis[acetonitrile]palladium(II). The Pd atom lies on a centre of symmetry and the compound is isomorphous with the platinum analogue.

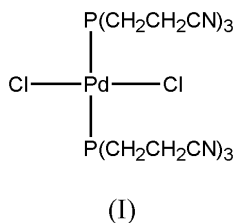
Received 24 May 2001

Accepted 4 June 2001

Online 15 June 2001

Comment

The use of tris(2-cyanoethyl)phosphine (CEP) as a ligand has attracted considerable interest since it has structural similarity to *n*-alkylphosphines and yet possesses different electronic properties (Cotton *et al.*, 1981). CEP has been utilized in the preparation of many metal complexes, including $[\text{HgCl}_2(\text{CEP})_n]$ (Bell *et al.*, 1984) and $\text{Ni}_4(\text{CO})_6(\text{CEP})_4$ (Bennett *et al.*, 1967). The simple dichloro-*trans*-bis(CEP)platinum(II) complex is known (Khan *et al.*, 1993), and as part of our continuing studies on transition metal CEP complexes, we decided to examine the palladium analogue dichloro-*trans*-bis[tris(2-cyanoethyl)phosphine]palladium(II), (I).



The title compound lies with the palladium on a centre of symmetry (Fig. 1) and is isomorphous with the platinum analogue. All molecular geometry parameters lie within the normal ranges (Allen *et al.*, 1987; Orpen *et al.*, 1989).

Experimental

Compound (I) was synthesized by the reaction of $\text{P}(\text{CH}_2\text{CH}_2\text{CN})_3$ with $[\text{PdCl}_2(\text{CH}_3\text{CN})_2]$ generated *in situ* by heating PdCl_2 with CH_3CN under reflux for 1.5 h. Two equivalents of $\text{P}(\text{CH}_2\text{CH}_2\text{CN})_3$ were added to the hot CH_3CN solution of $[\text{PdCl}_2(\text{CH}_3\text{CN})_2]$, causing an immediate colour change from orange to yellow. The reaction was heated under reflux for a further hour and filtered. The solvent was removed *in vacuo* from the filtrate and the sample crystallized from CH_3CN to give an 84% yield of (I) as a yellow crystalline solid. Crystals suitable for X-ray diffraction were obtained by slow evaporation from a CH_3CN solution.

Crystal data

[PdCl₂(C₉H₁₂N₃P)₂]
M_r = 563.67
 Orthorhombic, *Pbca*
a = 9.1308 (11) Å
b = 14.0077 (19) Å
c = 19.012 (3) Å
V = 2431.7 (6) Å³
Z = 4
D_x = 1.540 Mg m⁻³

Mo *K*α radiation
 Cell parameters from 22 reflections
 $\theta = 15.3\text{--}22.4^\circ$
 $\mu = 1.13\text{ mm}^{-1}$
T = 293 (2) K
 Block, yellow
 0.52 × 0.43 × 0.40 mm

Data collection

Nonius MACH-3 diffractometer
 ω -2 θ scans
 Absorption correction: ψ scan
 (North *et al.*, 1968)
T_{min} = 0.966, *T_{max}* = 1.000
 2665 measured reflections
 2256 independent reflections
 1853 reflections with *I* > 2σ(*I*)

R_{int} = 0.012
 $\theta_{\text{max}} = 25.5^\circ$
h = 0 → 11
k = 0 → 16
l = -22 → 22
 3 standard reflections
 frequency: 60 min
 intensity decay: 0.8%

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.027
wR (*F*²) = 0.076
S = 1.19
 2256 reflections
 133 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0248P)^2 + 3.6835P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.54\text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.48\text{ e \AA}^{-3}$

All H atoms were clearly defined in difference maps and were then treated as riding atoms with a C–H distance 0.95 Å and *U_{iso}*(H) = 1.2*U_{eq}*(C). Examination of the structure with *PLATON* (Spek, 2000) showed that there were no solvent-accessible voids in the crystal lattice.

Data collection: *CAD-4 Data Collection Software* (Nonius, 1998); cell refinement: *CAD-4 Data Collection Software*; data reduction: *maxUs* (Mackay *et al.*, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2000); software used to prepare material for publication: *SHELXL97* and *PLATON*.

We thank Enterprise Ireland for a studentship (MK) and Cork University Foundation for the purchase of an X-ray diffractometer.

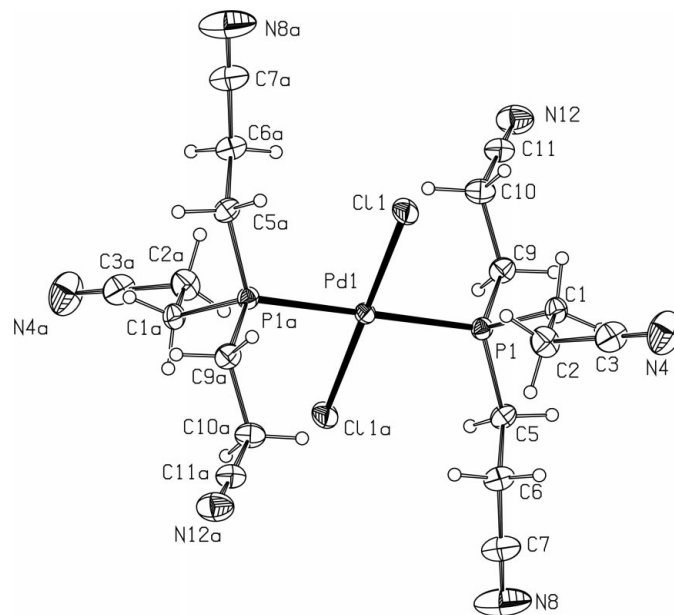


Figure 1

A view of the structure of (I) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
 Bell, N. A., Goldstein, M., March, L. A. & Nowell, I. W. (1984). *J. Chem. Soc. Dalton Trans.* pp. 1621–1624.
 Bennett, M. J., Cotton, F. A. & Winquist, B. H. C. (1967). *J. Am. Chem. Soc.* **89**, 5366–5372.
 Cotton, F. A., Darensbourg, D. J. & Ilsley, W. H. (1981). *Inorg. Chem.* **20**, 578–583.
 Khan, M. N. I., King, C., Fackler, J. P. & Winpenny, R. E. P. (1993). *Inorg. Chem.* **32**, 2502–2505.
 Mackay, S., Gilmore, C. J., Edwards, C., Tremayne, M., Stuart, N. & Shankland, K. (1998). *maxUs*. University of Glasgow, Scotland, Nonius BV, Delft, The Netherlands, and MacScience Co. Ltd, Yokohama, Japan.
 Nonius (1998). *CAD-4 Data Collection Software*. Nonius BV, Delft, The Netherlands.
 North, A. C. T., Phillips, D. C. & Matthews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
 Orpen, A. G., Brammer, L., Allen, F. H., Kennard, O., Watson, D. G. & Taylor, R. (1989). *J. Chem. Soc. Dalton Trans.* pp. S1–83.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Spek, A. L. (2000). *PLATON*. Version of May 2000. University of Utrecht, The Netherlands.